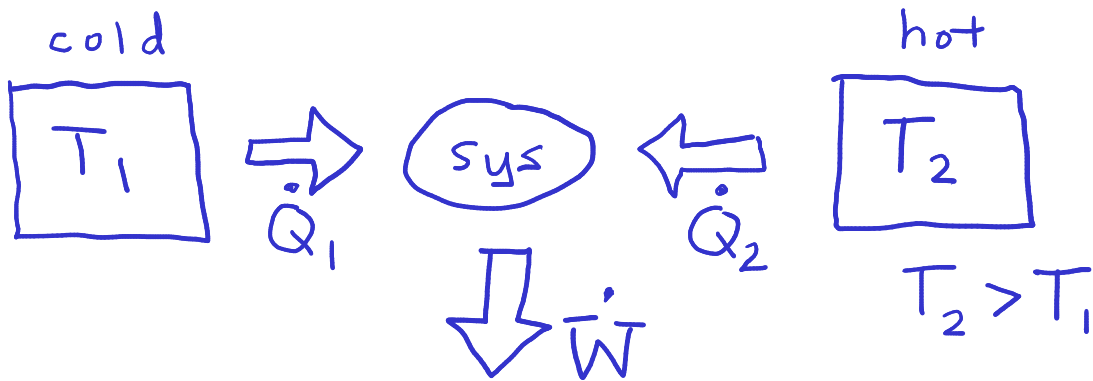


PHYS 414: 3-30-20



energy conservation: $\dot{Q}_1 + \dot{Q}_2 = \dot{E} + \dot{W}$

entropy budget: $\dot{S}^i = \dot{S} - \frac{\dot{Q}_1}{T_1} - \frac{\dot{Q}_2}{T_2} \geq 0$

stat. state: $\dot{S} = 0, \dot{E} = 0$

case 1: $\dot{Q}_2 > 0$ (hot bath donates heat)

$$\frac{\dot{W}}{\dot{Q}_2} = \eta_{\text{effic.}} = 1 - \frac{T_1}{T_2} - \frac{T_1 \dot{S}^i}{\dot{Q}_2}$$
$$\leq 1 - \frac{T_1}{T_2} \quad \text{Carnot inequality}$$

$$\Rightarrow \dot{W} \leq \dot{Q}_2$$

$$\dot{Q}_1 + \dot{Q}_2 = \dot{W}$$

$\dot{Q}_1 = \dot{W} - \dot{Q}_2 \leq 0$ (waste energy dumped into cold bath)

To achieve Carnot efficiency: $\dot{S}^i \rightarrow 0$

$$\dot{S}^i = \frac{k_B}{2} \sum_{nmk} J_{nm}^{(k)} \ln \frac{W_{nm}^{(k)} P_m}{W_{mn}^{(k)} P_n} \geq 0$$

only way to achieve $\dot{S}^i \rightarrow 0$

is to have all

currents $J_{nm}^{(k)} \rightarrow 0$

but then all dotted quantities $\rightarrow 0$

ie. $\dot{Q}_k = \frac{1}{2} \sum_{nmk} J_{nm}^{(k)} (E_n - E_m + V_{nm}^{(k)})$

$$\rightarrow 0$$

$$\dot{W} \rightarrow 0$$

this is a very boring limit:

approach Carnot effic.

as power output $\dot{W} \rightarrow 0$

Recall cyclic driving $W_{nm}(t) = W_{nm}(t + \tau)$

analogous formulation of above

derivation:

$\dot{A} \rightarrow$
replace

ΔA

difference
over one
cycle

$$\dot{E} = \dot{S} = 0 \rightarrow \Delta \bar{E} = \Delta \bar{S} = 0, \text{ etc.}$$

analogous Carnot bound:

work
over
one cycle

$$\rightarrow \frac{\Delta W}{\Delta Q_2} \leq 1 - \frac{T_1}{T_2}$$

applies
to
cyclic
engines

achieve Carnot effic

when $\Delta S_i = \int_0^\tau \dot{S}_i dt \rightarrow 0$

(very slow driving
 \Rightarrow "quasistatic")

$\tau \rightarrow \infty$ (cycle gets long)

$$\frac{\Delta W}{\tau} \rightarrow 0 \quad \text{mean power} \rightarrow 0$$



Interesting question:

Max. effic \iff Zero power

What is effic. at max. power?

Here no known universal bound.

System-specific?

Case 2: $\dot{Q}_1 > 0$ (cold bath gets colder)
refrigerator!

entropy budget
w/ $\dot{S} = 0$

$$\dot{S}^i = -\frac{\dot{Q}_1}{T_1} - \frac{\dot{Q}_2}{T_2}$$

$$\Rightarrow \dot{Q}_2 = -T_2 \left(\frac{\dot{Q}_1}{T_1} + \dot{S}^i \right) < 0$$

hot bath gets hotter
(machine is also a
"heat pump")

$$|\dot{Q}_2| > |\dot{Q}_1| \Rightarrow \dot{W} = \dot{Q}_1 + \dot{Q}_2 < 0$$

law of plugging in
the refrigerator
(ext. env. has to do
work on sys.)

refrig. coeff. $\eta_R = \frac{\dot{Q}_1}{-\dot{W}} = \frac{\text{heat rate out of cold bath}}{\text{power input}}$

$$= \frac{T_1}{T_2 - T_1} \left(1 + \frac{T_2 \dot{S}^i}{\dot{W}} \right)$$

real
refrig.
 $\eta_R \approx 4$

$$\eta_R \leq \frac{T_1}{T_2 - T_1}$$
$$\approx 11$$

$$T_2 = 298 \text{ K}$$
$$T_1 = 273 \text{ K}$$
$$\Rightarrow$$

heat pump coeff.

$$\eta_H = \frac{-\dot{Q}_2}{-\dot{W}} = \frac{\text{heat pump rate}}{\text{power input}}$$
$$= 1 + \eta_R$$