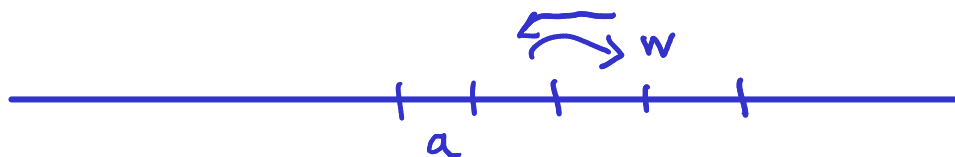


PHYS 320/420: Lecture 5

$$\text{MSD: } \langle \Delta_i^2 \rangle_t = a^2 (\langle i^2 \rangle_t - 2i_0 \langle i \rangle_t + i_0^2)$$

$$\Delta_i = a(i - i_0)$$



Ω_{ij} = trans. rate $j \rightarrow i$ ($i \neq j$)

$$\frac{d\langle i^n \rangle_t}{dt} = \sum_j \sum_{i \neq j} (i^n - j^n) \Omega_{ij} P_j$$

$$\Omega = \begin{pmatrix} -w & w & & & \\ w & -2w & w & & \\ & w & -2w & w & \\ & & & \ddots & \ddots \\ & & & & \ddots & \ddots \end{pmatrix} \begin{array}{l} \sum_i P_i(t) = 1 \\ \frac{d}{dt} \sum_i P_i(t) = 0 \end{array}$$

$$\Rightarrow n=1: \frac{d\langle i \rangle_t}{dt} = w P_1(t) - w P_N(t)$$

$$n=2: \frac{d\langle i^2 \rangle_t}{dt} = 2w \sum_{j=2}^{N-1} P_j(t) + 3w P_1(t) + (1-2N)w P_N(t)$$

assume N is large (a small) \downarrow
molecule is far from walls

$$P_1(t) \approx 0 \quad P_N(t) \approx 0$$

$$\Rightarrow 1 = \sum_{j=1}^N P_j(t) \approx \sum_{j=2}^{N-1} P_j(t)$$

$$\Rightarrow \frac{d\langle i \rangle_t}{dt} = 0, \quad \frac{d\langle i^2 \rangle_t}{dt} = 2w$$

no bias toward left or right

initial condition: all particles in our ensemble start i_0 at $t=0$:

$$\langle i \rangle_0 = i_0 \quad \langle i^2 \rangle_0 = i_0^2$$

$$\langle i \rangle_t = i_0 \quad \langle i^2 \rangle_t = 2wt + i_0^2$$

$$\langle \Delta_i^2 \rangle_t = 2wa^2t \quad \text{MSD}$$

$$\equiv 2Dt$$

$$D = wa^2 \quad \text{diffusion constant}$$

$$\text{units: } \frac{\text{length}^2}{\text{time}}$$

this result is for 1 dimension (\hat{x})
but by symmetry you would get
same answer for other axes

$$\vec{r} = a(i, j, k) \quad \langle \Delta_j^2 \rangle_t = 2Dt$$

$$\vec{r}_0 = a(i_0, j_0, k_0) \quad \langle \Delta_k^2 \rangle_t = 2Dt$$

$$\Delta_j = a(j - j_0)$$

$$\Delta_k = a(k - k_0)$$

$$\begin{aligned} 3D \text{ MSD} &: \langle (\vec{r} - \vec{r}_0)^2 \rangle_t \\ &= \langle a^2(i - i_0)^2 + a^2(j - j_0)^2 \\ &\quad + a^2(k - k_0)^2 \rangle \\ &= 6Dt \end{aligned}$$

rough timescale for diffusion to
cover a distance L in one dim:

$$\text{MSD} \sim L^2 = 2Dt \Rightarrow t \sim \frac{L^2}{2D}$$

Survey of diffusion coeff.
scales in biology:

Stokes law: sphere of radius R
in a liquid of
viscosity η at
temp. T

$$w a^2 = D = \frac{k_B T}{6\pi \eta R}$$

$$R = 1 \text{ nm}$$

$$D = 245 \text{ } \mu\text{m}^2/\text{s}$$

$$k_B = 1.38 \times 10^{-23} \frac{\text{J}}{\text{K}}$$

$$T = 298 \text{ K}$$

$$\eta = 0.89 \times 10^{-3}$$

$\text{Pa} \cdot \text{s}$

for water